

## Miller Levine Biology Chapter 2

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Section 2-1: The Nature of Matter The subatomic particles that make up atoms are protons, neutrons, and electrons. Because they have the same number of protons, all isotopes of an element have the same chemical properties. The main types of chemical bonds are covalent bonds and ionic bonds.

### Chapter 2 Resources - BIOLOGY by Miller & Levine

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### The Macaw Book - BIOLOGY by Miller & Levine

This 5 page reading guide covers chapter 8 section 1 and chapter 8 section 2 of the 2019 Miller and Levine Biology Textbook. Your students will interact with the textbook as they work through a variety of question styles, including short answer, matching, fill in the blank, and word scrambles.

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Chapter 2: The Chemistry of Life The Nature of Matter Atoms are the smallest, indivisible particles of matter First theorized by Democritus about 2500 years ago (ya) Atoms made of two parts that consist of a combination of three different subatomic particles Nucleus is the ?core? of an atom Protons have a positive charge Neutrons have a neutral charge Electron cloud Divided into orbitals (layers) First layer is smallest and contains maximum of 2 electrons Second layer contains maximum of 8 ...

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Academic Biology Classroom Lectures . The following is a link to your Academic Biology notes. Your text is Miller Levine 2010 Semester 1 – chapters are listed in the order they are covered. Quarter 1. Chapter 1 The Science of Biology and text pp 190-192 Microscopes. Chapter 2.1-2.2 The Chemistry of Life. Chapter 3 The Biosphere

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