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Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below. a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by 1°C ____ 2. SI unit of energy ____ 3.

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Chapter 17 Thermochemistry 183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

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chapter 17 thermochemistry worksheet answers SECTION 17.1
THE FLOW OF ENERGY HEAT AND WORK ... Chapter 17
Thermochemistry187 10 Complete the enthalpy diagram for the
combustion of natural gas Use the thermochemical equation in
the first paragraph on page 517 as a

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A camp fire is an example of basic thermochemistry. The
reaction is initiated by the application of heat from a match. The
reaction converting wood to carbon dioxide and water (among
other things) continues, releasing heat energy in the process.

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