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Worksheet - Chapter 3 A - Atomic Structure Atomic Theory Matching - Sec. 1 & 2 & 3 ____1.This device gave scientists their first A.Democritus. image of individual atoms. B. John Dalton ____2.He confirmed the existence of the . neutron.C. J.J. Thomson ____3.He developed the beginnings of the modernD. James Chadwick. atomic theory in the 1800s.E ...

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CHAPTER 5, Atomic Structure and the Periodic Table(continued) An atom of neon-22 has two more neutrons in its nucleus than an atom of neon-20. isotopes true b c a carbon-12; 12 amu false The values of atomic masses measured in grams are inconveniently small and impractical to work with. Guided Reading and Study Workbook43

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

CHAPTER 5, Atomic Structure and the Periodic Table(continued) Just How Small Is an Atom? (page 108) 6. Suppose you could grind a sample of the element copper into smaller and smaller particles. The smallest particle that could no longer be divided, yet still has the properties of copper, is ____ . 7.

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

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CHEMISTRY NOTES Chapter 5 Atomic Structure and the ...

Chapter 5 Atomic Structure. Franklin. Crookes. LaVoisier. John Dalton. discovered electricity in lightning storm. developed cathode ray tube. developed the law of conservation of mass, particles not creat.... combined Democritus' writings, the work of LaVoisier, crookes,....

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ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4 WORKSHEET PART A Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons. Isotope Symbol 131i S31 Us |He SAI 81W 36Kr SIDU 37KD Atomic Number Mass Number Protons Electrons Neutrons Isotope Name PARTB

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