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ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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12.3 Arrange the following compounds in increasing order of their boiling points:

$\text{CH}_3\text{CHO}$ ,  $\text{CH}_3\text{CH}_2\text{OH}$ ,  $\text{CH}_3\text{OCH}_3$ ,

$\text{CH}_3\text{CH}_2\text{CH}_3$  Ans. The order is :  $\text{CH}_3$

$\text{CH}_2\text{CH}_3 < \text{CH}_3\text{OCH}_3 < \text{CH}_3\text{CHO}$

$< \text{CH}_3\text{CH}_2\text{OH}$  All these compounds

have comparable molecular masses  $\text{CH}_3$

$\text{CH}_2\text{OH}$  undergoes extensive

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intermolecular H-bonding and thus its b.pt. is the highest.  $\text{CH}_3\text{CHO}$  is more polar than  $\text{CH}_3\text{OCH}_3$  so ...

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same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution.

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CHAPTER 12 REVIEW Solutions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at

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the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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