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molecular theory. ideal  
gas. elastic collision.  
expansion for gases.  
based on idea that  
particles of matter are  
always in motion.  
hypothetical gas that  
fulfills 5 assumptions.  
no net loss of total  
kinetic energy. no  
definite shape or  
volume, takes shape of  
container; accordi....

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ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left.

b ionic crystal (a) has mobile electrons in the crystal

c covalent molecular crystal (b) is hard, brittle, and nonconducting

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(better know as water)  
is the key to nearly  
everything. It falls from  
the sky, makes up 60%  
of our bodies, and ...  
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approximate  
equilibrium vapor  
pressure of each of the  
following at the  
specified temperature.  
a. water at 80C b.

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diethyl ether at 20°C  
ethanol at 60°C

**Using Figure 14,  
estimate the  
approximate  
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Calculations and  
equations 10, that can  
be made from 109 ×  
104 kilograms of

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phosphorus in the  
second of these three  
reactions The answer is  
 $250 \times 10^4$  kg P 40 10

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