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Cambridge End Of Chapter Test

10 Marking scheme: End-of-chapter test. 1 a Element Carbon Hydrogen Oxygen Percentage 62.1 10.3 27.6. Mass in 100 g 62.1 g 10.3 g 27.6 g. Number of moles. $12.0 \text{ g} / 62.1 \text{ g mol}^{-1} = 0.193$ $1.0 \text{ g} / 10.3 \text{ g mol}^{-1} = 0.097$ $16.0 \text{ g} / 27.6 \text{ g mol}^{-1} = 0.580$ [1] Relative number of atoms 1.73 5.18 = 3 1.73 10.3 = 6 1.73 1.73 = 1.

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13 Marking scheme: End-of-chapter test. 1 a. bC_nH_{2n+1}OH [1] cButan-1-ol and 2-methylpropan-2-ol [1] structural isomers [1] dCompounds with the same molecular formula [1] but different structural/displayed formulae [1] e2-methylbutan-2-ol [1] 2 aC₆H₁₂O₆ → 2CH₃CH₂OH + 2CO₂[1] fermentation [1]

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3 Marking scheme: End-of-chapter test COAS Biology 2 Teacher Resources Original material © Cambridge University Press 2009 2 5 a i Glucagon secretion increases in A ...

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